

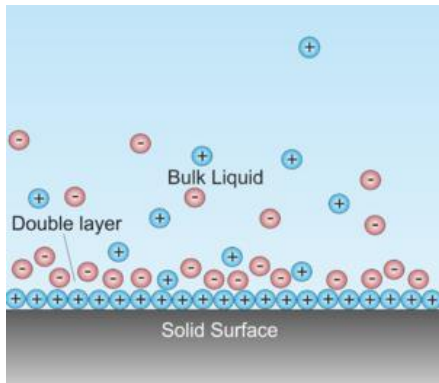
Molecular dynamics simulation at a constant electrode electrostatic potential

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2017 7/24 CREST Workshop

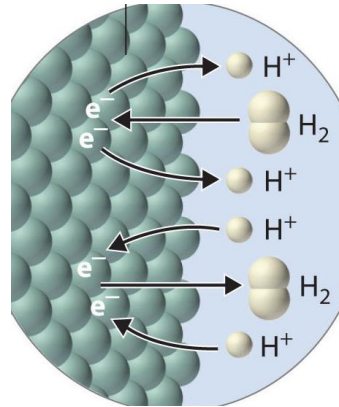
Electrochemistry

surface physics



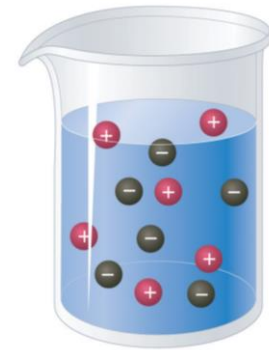
Wikipedia “Double layer (surface science)”

chemical reaction at electrode surface



“The Chemistry LibreTexts library” web page

electrolyte solution

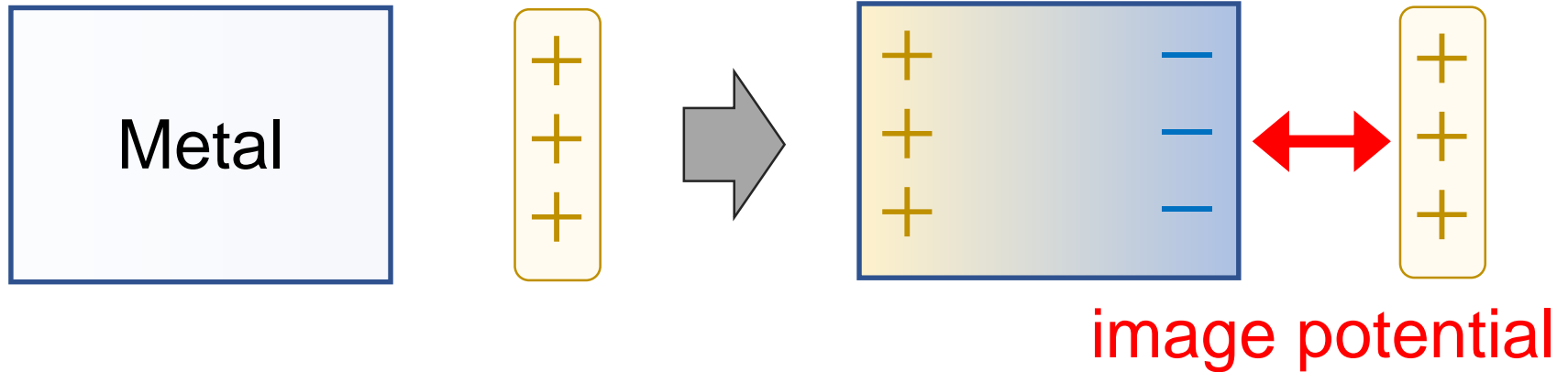


“Aqueous Solutions”, section 4.1 from the book [Principles of General Chemistry](#)

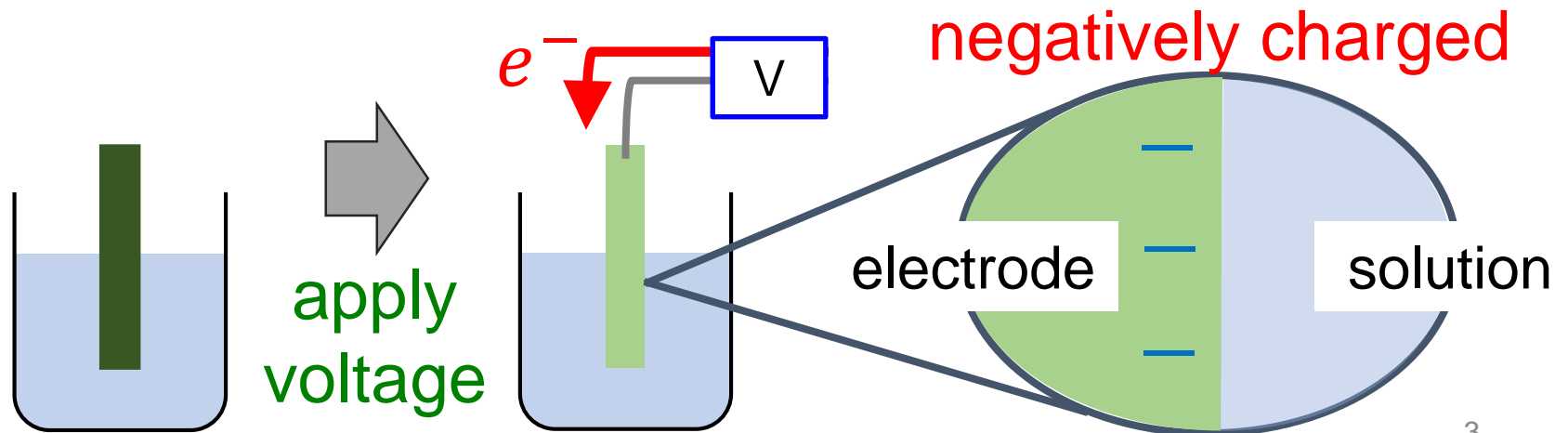
Battery, Capacitor, Electric sensor, Corrosion, ...

Metal Electrode Surface

1. Polarization of metal surface

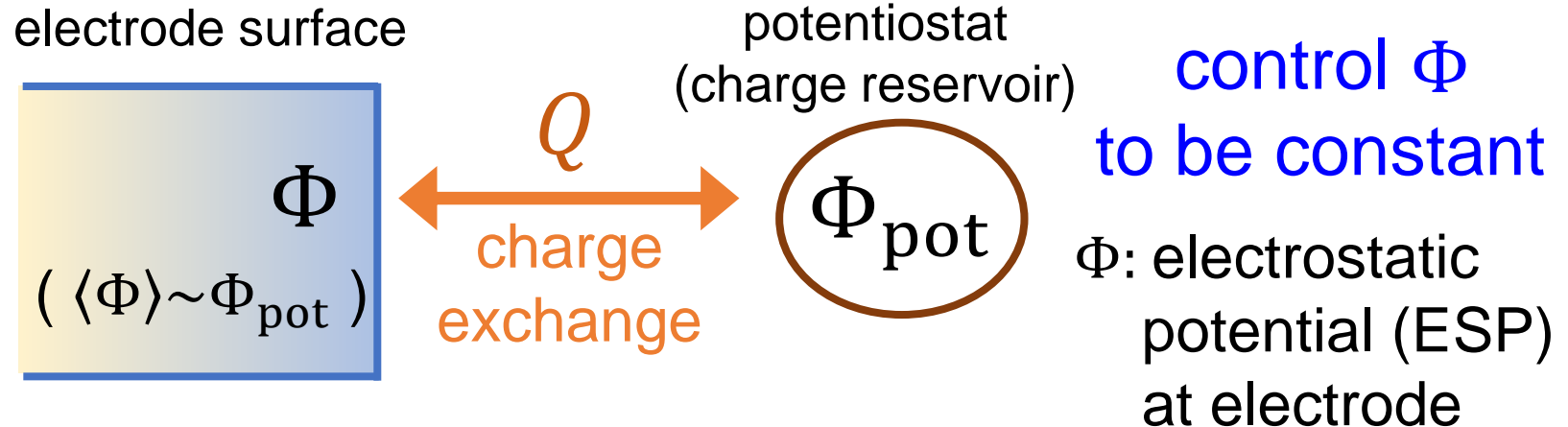


2. Electrode potential (voltage)



Modeling of Electrode Metal

➤ External potentiostat method



a kind of **extended system method**

cf. Andersen's constant pressure method

constant pressure (P, V) \longleftrightarrow external potentiostat (Φ, Q)

Siepmann and Sprik (*J. Chem. Phys.*, **102**, 511, 1995)

Reed and coworkers (*J. Chem. Phys.*, **126**, 084704, 2007)

Modeling of Electrode Metal

➤ Image charge method

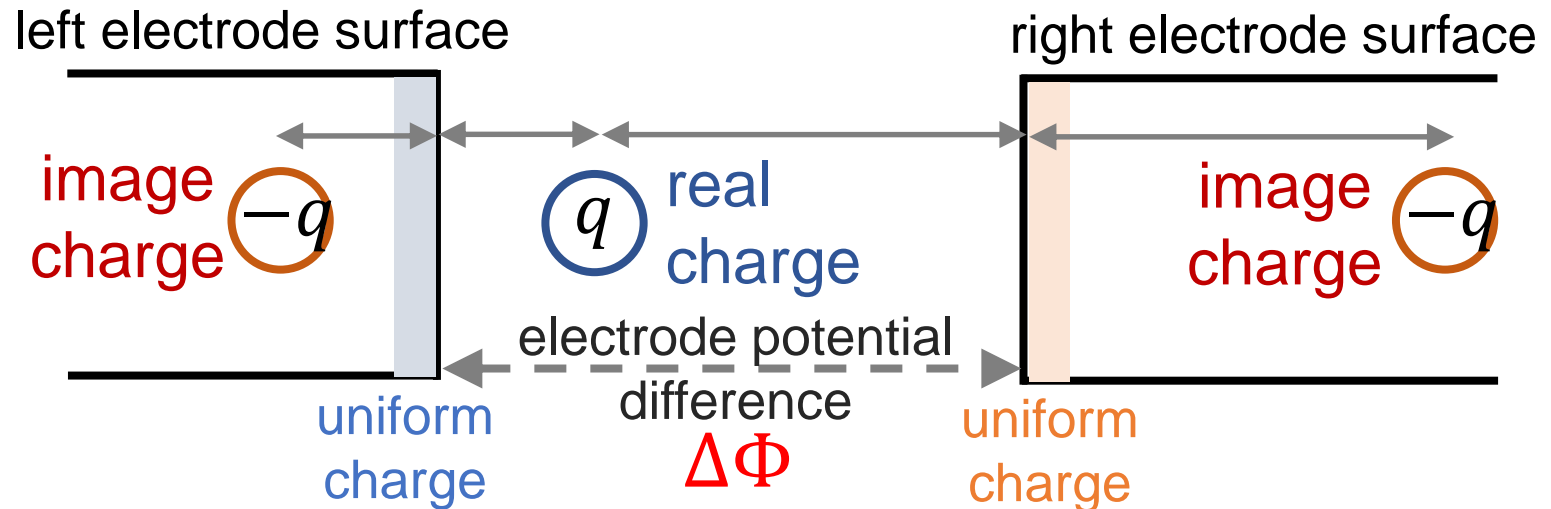


image charges in the interior of the electrode
⇒ **polarization** of electrode surface

uniform charges at the electrode surfaces
⇒ electrode **potential difference** $\Delta\Phi$

Petersen and coworkers (*J. Phys. Chem. C*, **117**, 3747, 2013)

Features of the two methods

External potentiostat method		Image charge method
✓	inhomogeneous surface	✗
✓	half cell model	✗
✓ ✗	computational cost	✓
✗	implementation	✓

Constant Electrode Potential Model

Hamiltonian for my external potentiostat method

$$H_{\text{NVE}\Phi}(\mathbf{r}, \mathbf{p}, \mathbf{Q}, \mathbf{P}_Q; \Phi_{\text{pot}}) = \sum_j^N \frac{\mathbf{p}_j}{2m_j} + \sum_i^n \frac{\mathbf{P}_{Q,i}}{2M_{Q,i}} + U(\mathbf{r}, \mathbf{Q}) - \mathbf{Q}\Phi_{\text{pot}} + E_{\text{self}}$$

\mathbf{Q} : electrode atomic point charges

$\mathbf{P}_Q, \mathbf{M}_Q$: fictitious momenta and masses associated with \mathbf{Q}

Φ_{pot} : potentiostat potential (parameter)

With this Hamiltonian, equations of motions for \mathbf{r} and \mathbf{Q} are constructed, and solved.

- ✓ 3D periodic boundary condition
- ✓ Point charge model (E_{self} is introduced for this model)
- ✓ Electrode charge as dynamical variable

Thermostated Electrode Charge

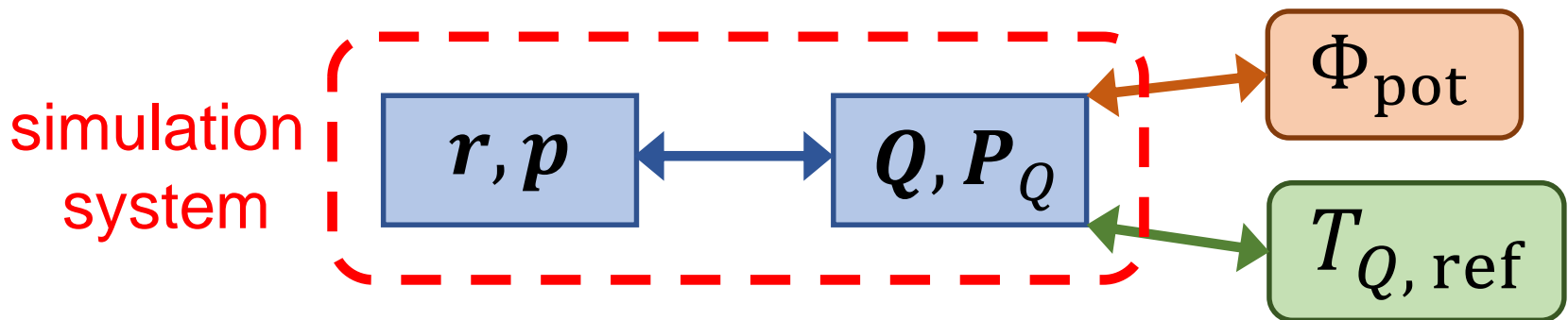
Electrode charges mimic electrons in the electrode.

~ Adiabaticity between atomic and charge dynamics should be achieved.



Thermostat for electrode charges should be attached.

$$H'_{\text{NVE}\Phi T_\Phi}(\Gamma) = \sum_j^N \frac{\mathbf{p}_j}{2m_j} + \sum_i^n \frac{\mathbf{P}_{Q,i}}{2M_{Q,i}} + \sum_k^{n_k} \frac{\mathbf{p}_{\xi,k}}{2M_{\xi,k}} + U'(r, Q) + kT_{Q,\text{ref}} \sum_k^{n_k} g_k \xi_k$$

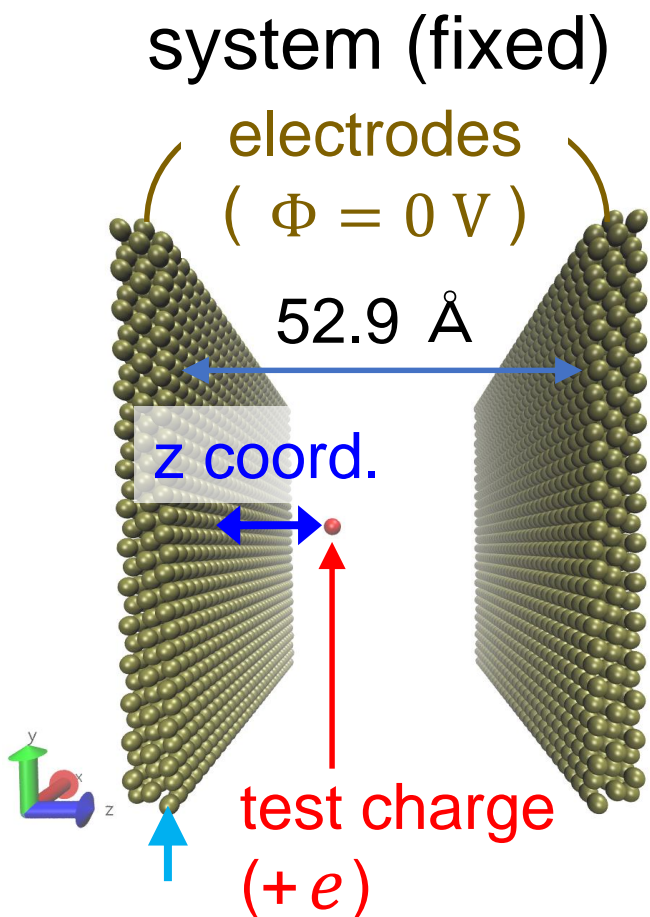


Computational Details

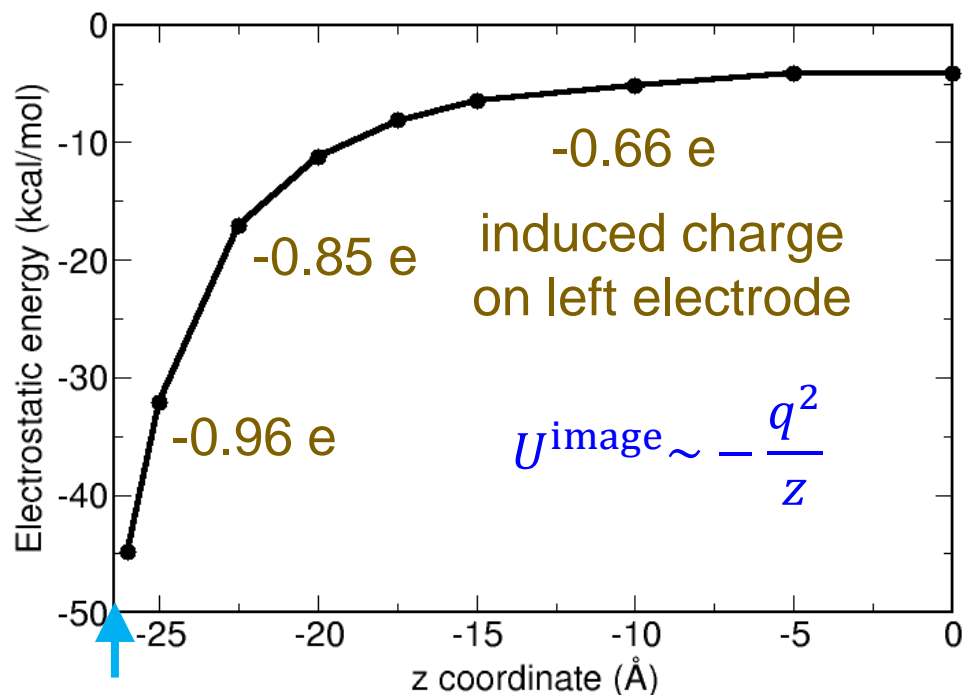
- Long-range electrostatic interaction:
Particle-Particle Particle Mesh Ewald method
- Ewald surface term:
included
- Electrode atom coordinates:
fixed
- MD program package:
modified LAMMPS program package

Other parameters and conditions are described below.

Polarization of Electrode

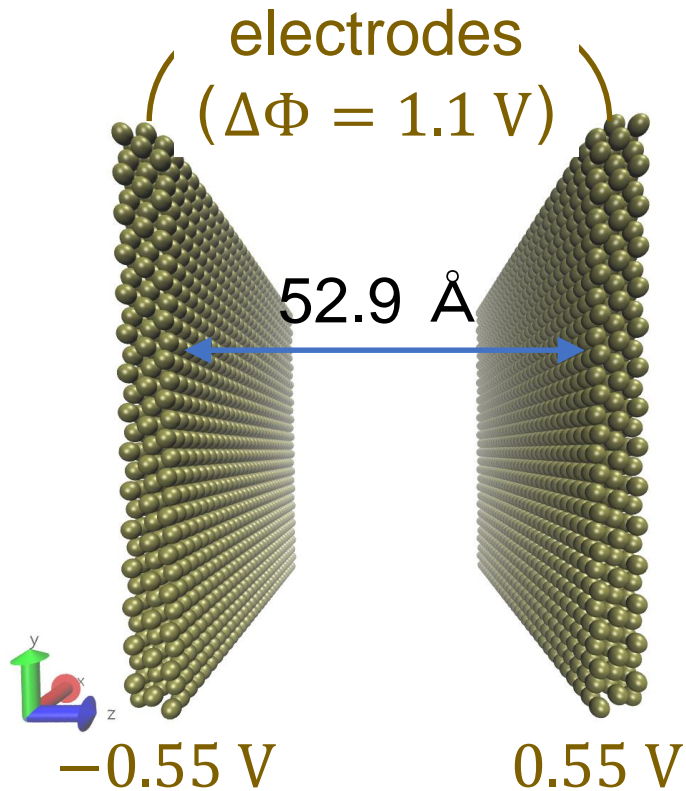


electrostatic energy
vs z coord. of a test charge



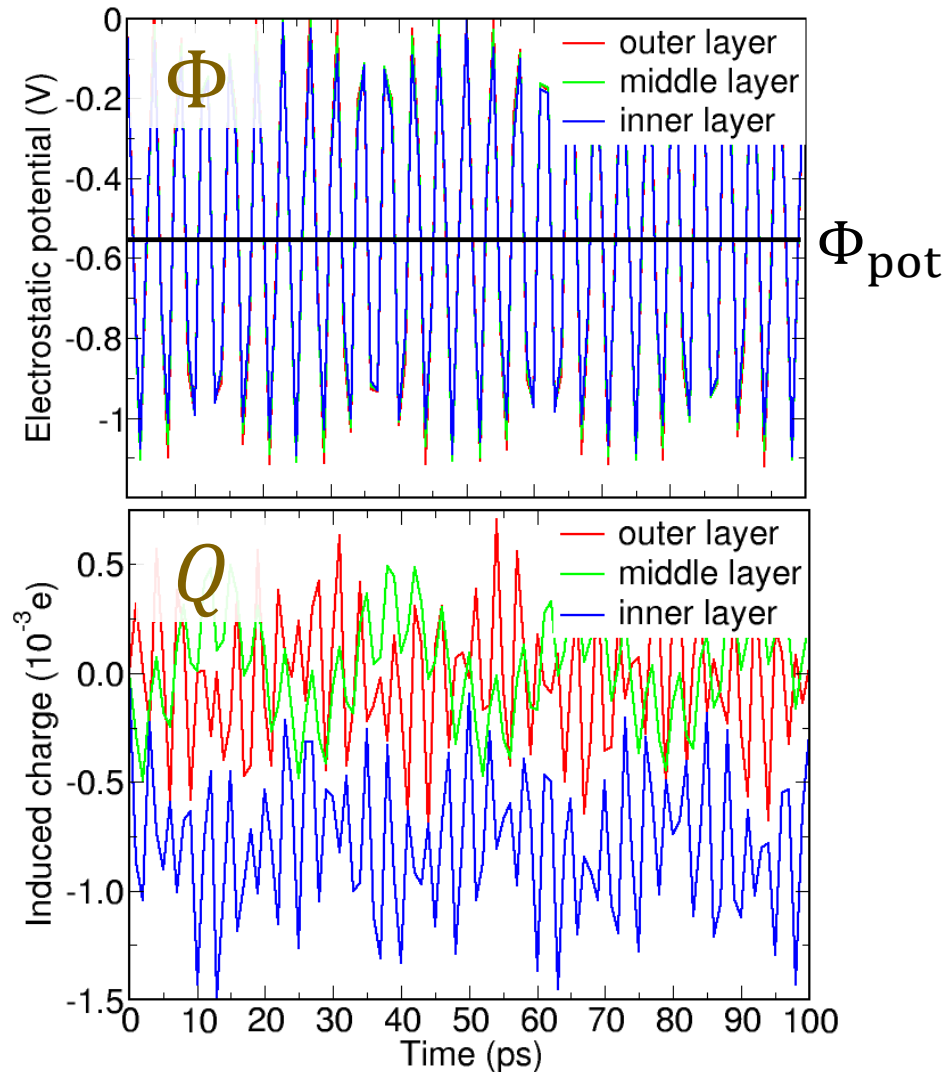
Electrodes are certainly polarized by a test charge.

Control of Electrode ES Potential

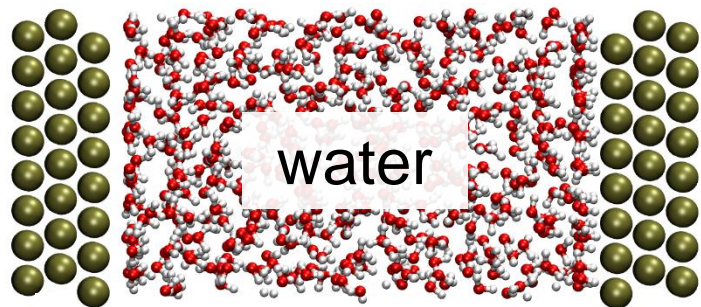


- ESP (Φ) is well controlled.
- Induced charges are mainly localized near the surface.

data from the left electrode

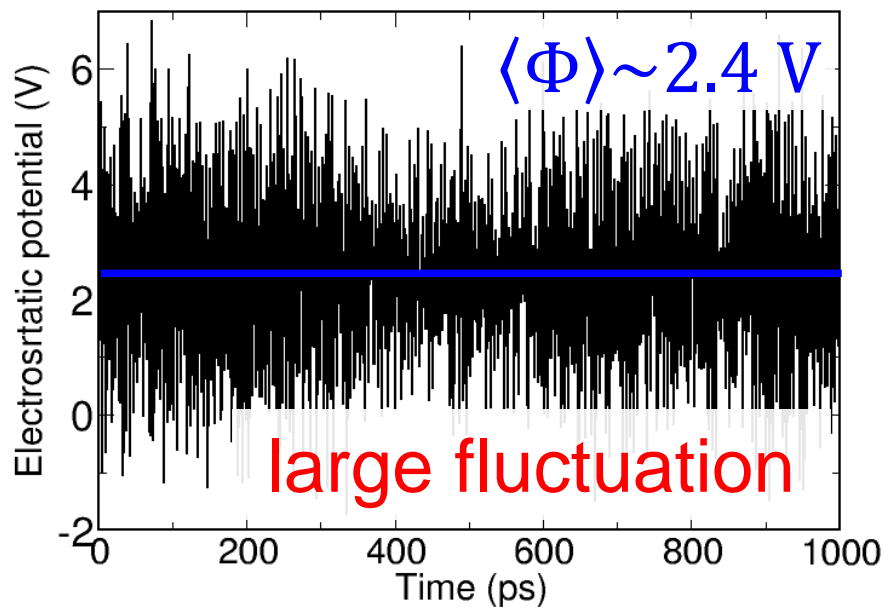


Adiabaticity between r and Q

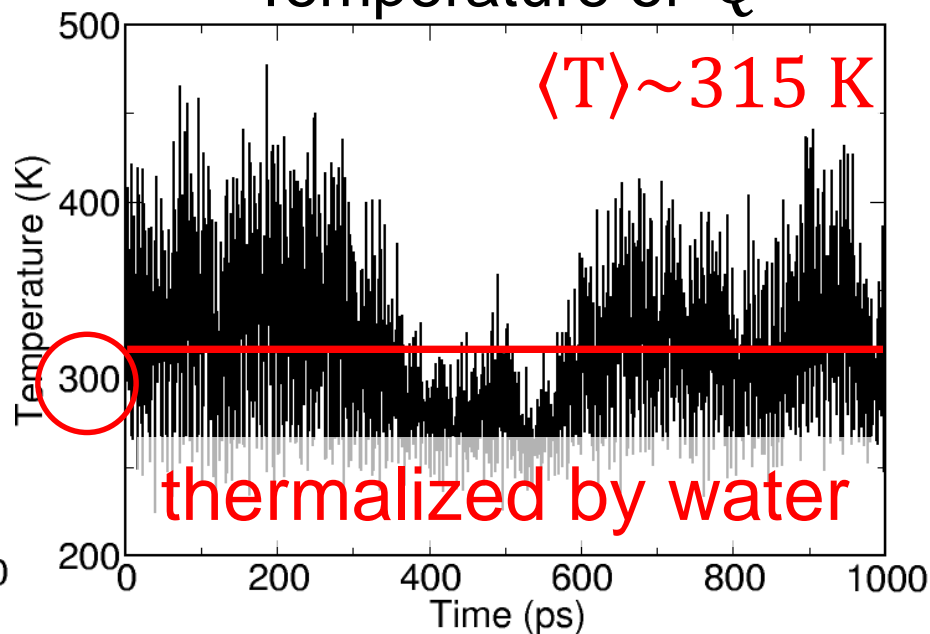


Φ_{pot} at left electrode: 2.4 V
temperature (water): 300 K

Φ at left electrode

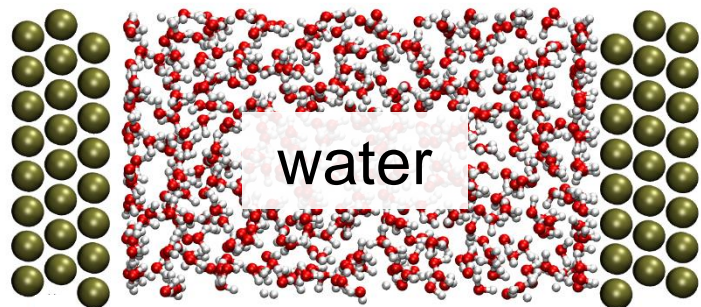


Temperature of Q



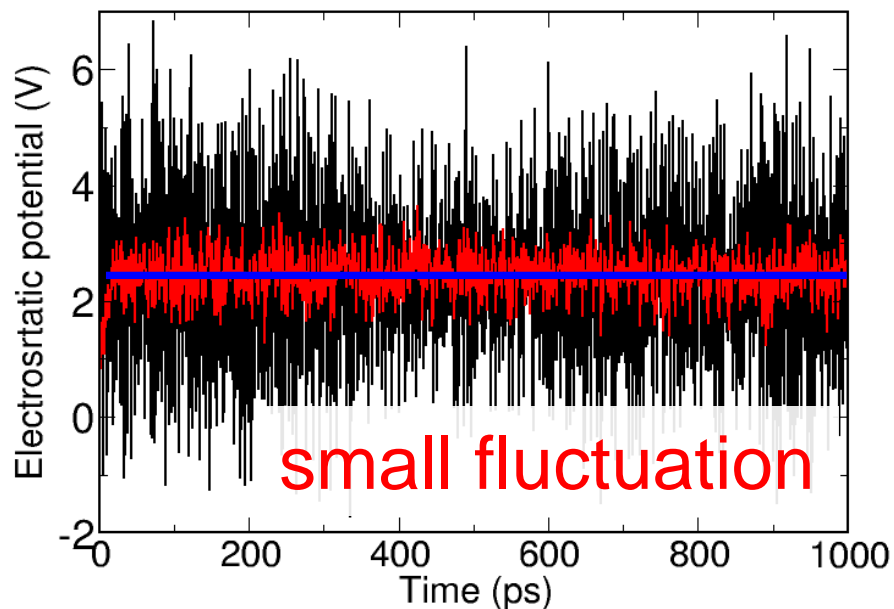
Heat flows from water to electrode charges.

Addition of Thermostat for Q

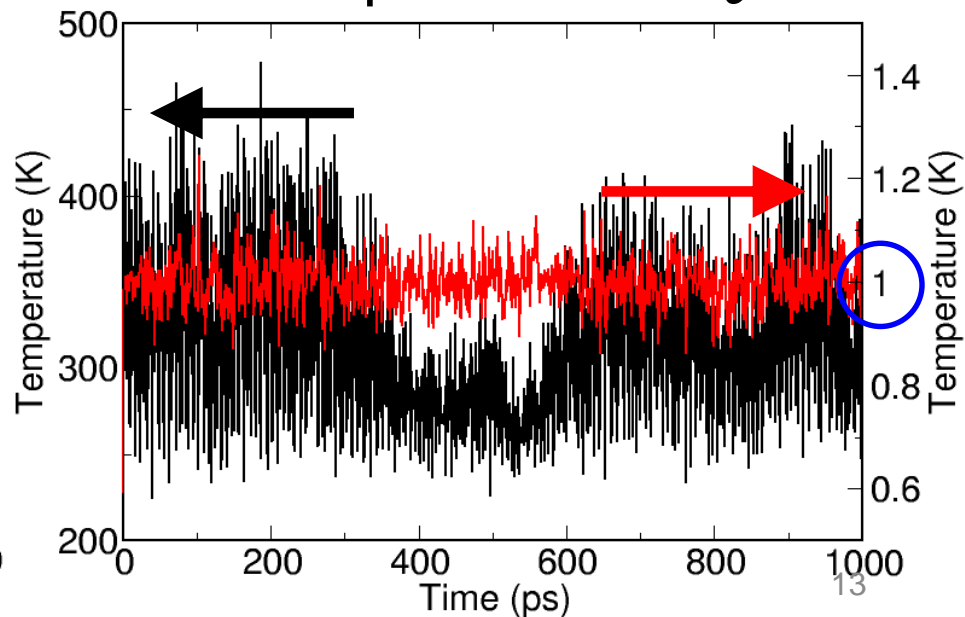


Φ_{pot} at left electrode: 2.4 V
temperature (water): 300 K
temperature (Q): 1 K

Φ at left electrode



Temperature of Q



Adiabaticity was achieved by thermostat for Q .

Summary

- New method accounting for the electrode polarization and ESP was developed.
 - ✓ Image charge and image potential can be calculated reasonably.
 - ✓ ESP on electrode was well controlled.
 - ✓ Adiabaticity was correctly achieved.
- Long time simulations which are infeasible in the first-principles MD are expected to be performed to investigate molecular behaviors near a surface in the more realistic level.